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Dimensionless numbers describe ratios of important fluid characteristics such as density, viscosity, speed of sound, and floAI-enhanced title and describes 12 major dimensionless numbers used in fluid mechanics and heat transfer analyses. It provides the symbol, formula, and significance For Later100%100% found this document useful, undefined100%(1)100% found this document useful u numbers used in fluid mechanics and heat transfer analyses. It provides the symbol, formula, and significance of each number, including the Reynolds, Prandtl, Schmidt, Lewis, Peclet, Stanton, Nusselt (or Sherwood), Grashof, Biot, Rayleigh, and Fourier numbers. The numbers are ratios that characterize different flow regimes, transport phenomena or compare relative magnitudes of forces and transport mechanisms. 100% (1)100% found this document useful (1 vote)457 views5 pages The document defines and describes 12 major dimensionless numbers used in fluid mechanics and heat transfer analyses. It provides the symbol, formula, and significance of each number, includAI-enhanced title and description 1. Important Dimensionless Numbers Keerthi Vasan MChemical Engineering Graduate (2020) kvasan 166@gmail.com Tamil Nadu, INDIA 1. Schmidt number (Sc): Sc gives the relative ease of momentum and mass transport in a flow system and it is the mass transfer analogue of Prandtl number (Pr). It relates the thickness of the momentum and diffusion boundary layers.Sc = Momentum diffusivity ()Mass diffusivity (D) = D = DD - Diffusivity of the system (m2/s), - Density (kg/m3), - Dynamic viscosity of the fluid(Pa s).2. Prandtl number (Pr): Pr gives the relative ease of momentum and energy transportin a flow system. It is more frequently used in heat transfer calculations. It relates thethickness of the momentum and thermal boundary layer. If Pr 1, heat diffusivity ()= kCP = CP k - Density (kg/m3), -Dynamic viscosity of the fluid (Pa s), k - Thermal conductivity(W/m K), CP - Specific heat capacity (J/kg K).3. Lewis number (Le): Le is used to characterize a flow system where simultaneous heatand mass transfer occurs. For a air-water vapour system, Le = 1. Which results in the Lewisrelation (used to prove that Wet Bult Temperature (TWBT) is equal to Adiabatic SaturationTemperature (Tas)).Le =ScPr=Thermal diffusivity ()Mass diffusivity (D)=kCP D=k D CPLewis relation derivation for air-water vapour system : Chilton - Colburn Analogy givesfF2|{z}jM=kC Sc2/3v| {z}jD=h Pr2/3G CP| {z}jH(1)Le = 1 implies, Pr = Sc. Therefore equation (1) gives, kCv= hG CP. Now by doing the following substitution = c MB, ky = kc c, Cp = Cs and kY = MB ky, we get Lewis relationashkY Cs = 1 . - Density (g/m3), k - Thermal conductivity (W/m K), CP - Specific heat capacity (J/kgK), fF - Fanning friction factor, kC - Mass transfer co-efficient (W/m2K), G - Mass transfer co-efficient for concentration driving force, v - fluid velocity (m/s), h - Convective heat transfer co-efficient (W/m2K), G - Mass transfer co-efficient for concentration driving force, v - fluid velocity (m/s), h - Convective heat transfer co-efficient (W/m2K), G - Mass transfer co-efficient for concentration driving force, v - fluid velocity (m/s), h - Convective heat transfer co-efficient (W/m2K), G - Mass transfer co-efficient for concentration driving force, v - fluid velocity (m/s), h - Convective heat transfer co-efficient (W/m2K), G - Mass transfer co-efficient for concentration driving force, v - fluid velocity (m/s), h - Convective heat transfer co-efficient for concentration driving force, v - fluid velocity (m/s), h - Convective heat transfer co-efficient for concentration driving force, v - fluid velocity (m/s), h - Convective heat transfer co-efficient for concentration driving force, v - fluid velocity (m/s), h - Convective heat transfer co-efficient for concentration driving force, v - fluid velocity (m/s), h - Convective heat transfer co-efficient for concentration driving force, v - fluid velocity (m/s), h - Convective heat transfer co-efficient for concentration driving force, v - fluid velocity (m/s), h - Convective heat transfer co-efficient force for concentration driving force for co Massvelocity in kg/ m2s (= v), Cs - Humid heat of the air-water vapour system, MB - Molecularweight of dry air (= 29 g/mol), c - Concentration of vapour in the mixture (mol/m3), ky -Mass transfer co-efficient for mole fraction driving force, kY - Mass transfer co-efficient for mole fraction driving force, kY - Mass transfer co-efficient for mole fraction driving force, kY - Mass transfer co-efficient for mole fraction driving force, kY - Mass transfer co-efficient for mole fraction driving force, kY - Mass transfer co-efficient for mole fraction driving force, kY - Mass transfer co-efficient for mole fraction driving force, kY - Mass transfer co-efficient for mole fraction driving force, kY - Mass transfer co-efficient for mole fraction driving force, kY - Mass transfer co-efficient for mole fraction driving force, kY - Mass transfer co-efficient for mole fraction driving force, kY - Mass transfer co-efficient for mole fraction driving force, kY - Mass transfer co-efficient for mole fraction driving force, kY - Mass transfer co-efficient for mole fraction driving force, kY - Mass transfer co-efficient for mole fraction driving force, kY - Mass transfer co-efficient for mole fraction driving force, kY - Mass transfer co-efficient for mole fraction driving force, kY - Mass transfer co-efficient for mole fraction driving force, kY - Mass transfer co-efficient for mole fraction driving force, kY - Mass transfer co-efficient for mole fraction driving force, kY - Mass transfer co-efficient for mole fraction driving force, kY - Mass transfer co-efficient for mole fraction driving force, kY - Mass transfer co-efficient for mole fraction driving force, kY - Mass transfer co-efficient for mole fraction driving force, kY - Mass transfer co-efficient for mole fraction driving force, kY - Mass transfer co-efficient for mole fraction driving force, kY - Mass transfer co-efficient for mole fraction driving force, kY - Mass transfer co-efficient for mole fraction driving force, kY - Mass transfer co-efficient for mole fraction convective heat transfer calculations. It is the ratio ofthermal energy convected to the fluid to the thermal energy conducted within the fluid. Inpractical applications, Pe is also known as Bodenstein number(Bo). Pe = Heat transport by convectionHeat transport by conduction = Re Pr = v L = L v Similarly mass transfer analogue of Pe is defined as Pe = Mass transport by convection was transport by conduction = Re Sc = v L D = L vDDispersion number is defined as 1Pe. Dispersion number is defined as 1Pe. Dispersio Dispersion numberIdealCSTRInfinite axialmixingVery high D Pe = 0 (or) Dispersion number = IdealPFRZero axialmixingVery low D or zero D Pe = (or) Dispersion number = IdealPFRZero axialmixingVery low D or zero D Pe = (or) Dispersion number = 0 - Density (g/m3), v - Fluid velocity (m/s), L - Characteristic length of the system (m), - Dynamic viscosity of the fluid (Pa s), - Momentum diffusivity (m2/s), - Thermaldiffusivity (m2/s), D - Diffusivity of the system (m2/s) / Dispersion co-efficient. 5. Nusselt number (Nu): It is defined as a ratio of convective heat flux to conductive heat flux to co fluxConductive heat flux (Fouriers law)=h 4Tk 4TL =h LkNu =1, Heat transfer across the fluid layer is by pure conduction 10, Laminar flow100 1000, More active convection or Turbulent flowh - Convective heat transfer co-efficient (W/m2K), T - Temperature (K), k - Thermalconductivity (W/m K), L - Characteristic length of the system (m).6. Sherwood number (Sh): It is the mass transfer analogue of Nusselt number (Nu). It isdefined as Sh = Convective mass flux (Ficks law) = kc 4CD 4CL = kc LDkc - Mass transfer co-efficient for concentration driving force, D - Diffusivity of the system (m).7. Biot number (Bi): It is similar too. Nusselt number (Nu) but it is for a solid body. Whereas, Nu is for a fluid layer. It arises in the transient heat conductive heat flux (Fouriers law)=Internal conductive resistance=Lk 1h = h Lk2 3. If Bi 0.1, then internal conductive heat flux (Fouriers law)=Internal conductive resistance flux (Fouriers law)=Internal conductive heat f resistance is zero which makes the system a perfectlumped heat capacitance system. There wont be any resistance for heat flow on thebody. So, temperature is uniform through out the body. If Bi 1, then internal conductive resistance is significant. So, temperature distributionis non-uniform across the body and the calculations become complex.Important fact: Ratio of Nusselt number to Biot number gives, NuBi=h Lk fluidh Lk solid=ksolidkfluid=RfluidRsolid=Conductive resistance of fluidConductive resistance of solidh - Convective heat transfer co-efficient (W/m2K), T - Temperature (K), k -Thermalconductivity (W/m K), L - Characteristic length of the system (m).8. Fourier number (Fo): It arises in the unsteady state heat conduction calculations. Mainlyin the lumped heat capacitance systems (from the term (hAt/CV)).hAtCV=htCL=hLk ktCL2 = Bi tL2 = Bi FoALh - Convective heat transfer co-efficient (W/m2K), A - Cross sectional area of the body(m2), t - Time (s), - Density of the body (g/m3), C - Specific heat capacity of the body (g/m3), C - Specific heat capacity of the body (m3), L - Characteristic length of the system (m), k - Thermal conductivity (W/m K), Bi - Biot number, - Thermal diffusivity (m2/s).9. Stanton number (St): St is generally used in the forced convective heat transfer calculations and in heat, mass and momentum transfer analogies like Reynolds analogy, Chilton- Colburn analogy, etc. It is defined as a ratio of heat transferred to a fluid to the heat ca-pacity of the fluid. The stanton number arises in the consideration of geometeric similarity of the momentum boundary layer, where it can be used to express a relationship betweenthe shear force at the wall (due to viscous drag) and the total heat transfer at the wall (due to thermal diffusivity). St = Heat transfer analogue of St is defined as St = ShRe Sc=kc LD v L D = kcvNu - Nusselt number, Re - Reynolds number, Pr - Prandtl number, Sh - Sherwood number, Sc - Schmidt number, h - Convective heat transfer co-efficient (W/m2K), - Density (g/m3), v - Fluid velocity (m/s), CP - Specific heat capacity (J/kg K), G - Mass velocity in kg/ m2s (= v), kc - Mass transfer co-efficient for concentration driving force, L - Characteristiclength of the system (m), D - Diffusivity of the system (m2/s), - Dynamic viscosity of thefluid (Pa s).10. Damkohler number (Da): It is used to relate the chemical reaction a general nthorder chemical reaction A B,Da is defined asDa =Rate of consumption of A by reactionRate of transport of A by convection=rA VFA0=kCnA0V0CA0=kCn1A0 3 4. Therefore for a 1storder chemical reaction, Da = k.rA - Rate of consumption of A (mol/m3), V - Volume of the reactionmixture (m3), FA0 - Initial volumeteric flow rate (m3/s), -Space/residence time (s) (= V0).11. Dean number (De): It is used in the study of flow and heat transfer in coiled tube/pipesand channels. De = RerD2RCDean numberReynolds number (Re): Boundary layer seperation. Re = Interial forceViscous force=maA=muA= A u2 uL A= u L=u LQ = Auut uL dudym - Mass (g), a - Acceleration (m/s2), - Shear stress (Pa), A - Area (m2), u - Velocity(m/s), t - Time (s), - Dynamic viscosity (m2/s).13. Froude number (Fr): Fr =sInterial forceGravity force=s A u2m g=s A u2 A L g=uLg V ' L3' ALm - Mass (g), A - Area (m2), u - Velocity (m/s), L - Characteristic length of the system(m), - Density (g/m3), g - Acceleration due to gravity (m/s2).14. Euler number = 1Ru= 1uq4P. For frictionless flow, Eu = 1. Euler number (Eu) is called Cavitation number (Ca) when 4P = P Pv.Pv- vapour pressure (Pa), A - Area (m2), u - Velocity (m/s), - Density (g/m3).15. Weber number (We) :We =rInertial forceSurface tension (N/m), L - Characteristic length of the system (m).4 5. 16. Mach number (Ma) :Mach number =rInertial forceElastic force=rAu2kA= urk=uC - Density (g/m3), A - Area (m2), u - Velocity (m/s), k - Elastic stress, C - Velocity of thesound in the fluid (m/s) (=qk).17. Grashof number (Gr): Gr is used in the free/natural convection calculations to classifythe flow (flow classification is done based on the value of the term GrPr). Gr has the samefunctionality as the Reynolds number (Re). Grashof number (Gr) is defined asGr = Buoyancy forceViscous force=FbFv(2)Since in natural convections, we need to consider Changes in density of the fluid whichresults in fluid motion. To account those density changes, we consider Thermal coefficient of expansion (). = 1V dVdT P = d(1)dT = 1ddT = $d(\ln d) = TTO(\ln d) =$ we get viscous force = interial force. Therefore, uL A = Au2 u = uL(8) Substituting u from equations (7), (9) and (2), we get Gr = (04T)(AL)g 2AL2 = 2L3g4T2(10) Property values (CP, , , , etc.) are measured at film temperature Tf = Ts+T02.5 6. Important relationship for natural and forced convection obtained by dimensional anal-ysis includes, Natural convection: Nu = f(Re, Pr) forced convection dominates T. Temperature (K), T0 - Free stream temperature (K), 0 - Bulk fluid density (g/m3), -Fluid density inside heated air (g/m3), P - Pressure (Pa), V - Volume (m3), n - Number of moles (moles), R - Universal gas constant (J/mol K), T - Temperature (K), mf - Mass offluid (g), g - Acceleration due to gravity (m/s2), Vf - Volume of fluid (m3), A - Area (m2), L - Characteristic length of the system (m), -Dynamic viscosity of the fluid (Pa s), u -Fluid velocity (m/s), CP - Specific heat capacity (J/kg K), Tf - Film temperature (K), Nu - Nusselt number, Re - Reynolds number, Re -}1Heat diffusion rate = = CPk=2L3g4TCPk Property values (CP, , , , etc.) are measured at film temperature Tf = Ts+T02. Ra (= GrPr) is used in the free convection calculations to classify the flow. - Density (g/m3), - Thermal coefficient of expansion (K1), T - Temperature (K), g -Acceleration due to gravity (m/s2), L - Characteristic length of the system (m), - Dynamic viscosity of the fluid (Pa s), CP - Specific heat capacity (J/kg K), Tf - Film temperature (K), Ts - Surface/wall temperature (K), To - Free stream temperature (and also it correlates thermally developing flow. Gz = Thermal capacityConvective heat flux = DHL ReScDH - Hydraulic diameter (m), L - Characteristic length of the system (m), Re - Revnoldsnumber, Pr - Prandtl number, Sc - Schmidt number, Pe - Peclet number, 20. Knudsen number (Kn): It is defined as the ratio of the molecular mean free path length of the system=L6 7. If Kn 10, Knudsen diffusion happens. Knudsen diffusivity (Dk) = DV3= 48.5DqTMKnudsen diffusion flux can be found by replacing DAB by Dk in the ficks law ofdiffusion.Dk 6= f(P)= f(Pore diameter, molecular weight) Kn 1100, Molecular weight Kn 1100, Transition region diffusion happens. If 10 Kn 1100, Transition region diffusion happens. If 10 Kn 1100, Transition region diffusion happens. Characteristic length of the system(m), - Dynamic viscosity of the fluid (Pa s), - Pore diameter(m), DAB - Diffusion coefficient (m2/s), P - Pressure (Pa), Important Correlations Using Dimensionless Numbers 1. Dittus-Boelter equation :Dittus-Boelter equation :Dittus-Boel Boelter equation for turbulent flow inside a circular cross sectional area pipe isNu = 0.023 Re0.8Prn(11)hdk= 0.023 vd0.8 CP kn(12)Where, n = (0.4, Fluid heated 0.3, Fluid cooled Similarly mass transfer analogue of Dittus-Boelter equation is Sh = 0.023 Re0.8Scn. Validitychecks to use the Dittus-Boelter equation are Ld 60, 0.6 Pr 100, and 2500 Re 1.25 106.Constant average velocity (v): From equation (12) we get, hd d0.8 h d0.2 Constant mass flow rate (m): In equation (12), replace v by QA v 1d2. We get, hd d0.8 h d0.2 Constant mass flow rate (m): In equation (12), replace v by QA v 1d2. We get, hd d0.8 h d0.2 Constant mass flow rate (m): In equation (12), replace v by QA v 1d2. We get, hd d0.8 h d0.2 Constant mass flow rate (m): In equation (12), replace v by QA v 1d2. We get, hd d0.8 h d0.2 Constant mass flow rate (m): In equation (12), replace v by QA v 1d2. We get, hd d0.8 h d0.2 Constant mass flow rate (m): In equation (12), replace v by QA v 1d2. We get, hd d0.8 h d0.2 Constant mass flow rate (m): In equation (12), replace v by QA v 1d2. We get, hd d0.8 h d0.2 Constant mass flow rate (m): In equation (12), replace v by QA v 1d2. We get, hd d0.8 h d0.2 Constant mass flow rate (m): In equation (12), replace v by QA v 1d2. 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We get, hd d0.8 h Diameter(m), k - Thermal conductivity (W/m K), - Density (g/m3), v - Average velocity (m/s), - Dynamic viscosity of the fluid (Pa s), CP - Specific heat capacity (J/kg K), L - Characteristiclength of the system (m), Q - Volumetric flowrate (m3/s), A - Area (m2).2. Sieder-Tate equation : Sieder-Tate equation for turbulent flow inside a circular cross sectional area pipe isNu = 0.027 Re0.8Pr(1/3) bw0.14(13)hdk = 0.027 vd0.8 CP k(1/3) bw0.14(15)Nu - Nusselt number, Re - Reynolds number, Pr -Prandtl number, h - Convective heattransfer co-efficient (W/m2K), d - Diameter (m), k - Thermal conductivity (W/m K), -Density (g/m3), v - Average velocity (m/s), - Dynamic viscosity of the fluid (Pa s), CP -Specific heat capacity (J/kg K), L - Characteristic length of the system (m), b - Dynamic viscosity of the fluid at bulk temperature (Pa s), w - Dynamic viscosity of the fluid at walltemperature (Pa s).3. Liquids metals (Hq, Na, Lead Bismuth alloy, etc): High liquid metals can be used where high heat removal is to be achieved (e.g.: Coolants in a nuclear reactors). Since for liquid metals can be used where high heat removal is to be achieved (e.g.: Thermal diffusivity (m2/s), Hg - Mercury, Na - Sodium, - Momentum diffusivity (m2/s), Pr - Prandtl number, Nu - Nusselt number, Nu - Nusselt number, Re - Peclet number, A. Mass transfer in liquids flowing past single sphereSh = kCdDAB = 2 + 0.95 Re(1/2)Sc(1/3)When sphere is placed in a still fluid, Re = 0. So, Sh = 2 .Sh - Sherwood number, kC - Mass transfer co-efficient for concentration driving force, d - Diameter (m), DAB - Diffusivity of the system (m2/s), Re - Reynolds number, Sc - Schmidtnumber. 5. For a convective mass transport over (external flow) a flat plate under laminar conditionSh = 0.664 Re(1/2)Sc(1/3) and local Shx = Sh2Sh - Sherwood number, Re - Reynolds number, Sc - Schmidt number. 8 The paper explores dimensionless groups in fluid mechanics, providing a comprehensive overview of various dimensionless numbers crucial for understanding fluid behavior under different conditions. It discusses significant groups such as Archimedes, Bingham, Bond, and Cauchy numbers, detailing their formulas, notations, and applications in practical scenarios involving fluid dynamics and flows. The analysis emphasizes the importance of these groups in predicting fluid behavior in various engineering and scientific applications.

Dimensionless numbers in fluid mechanics and their significance. Dimensionless numbers in fluid mechanics. Fluid dimensionless numbers.

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